

# Band Structure

Continued

# Bloch theorem

## Bloch Theorem

$$\Psi_{\vec{k}}(\vec{r}) = e^{i\vec{k}\cdot\vec{r}} u_{\vec{k}}(\vec{r})$$

$$u_{\vec{k}}(\vec{r}) = u_{\vec{k}}(\vec{r} + \vec{R})$$

$$T_{\vec{n}} \Psi_{\vec{k}}(\vec{r}) = \Psi_{\vec{k}}(\vec{r} + \vec{R}_{\vec{n}}) = e^{i\vec{k}\cdot\vec{R}_{\vec{n}}} \Psi_{\vec{k}}$$

*The eigenstates of a periodic one-electron Hamiltonian can be chosen to have the form of a plane wave times a function with the periodicity of the Hamiltonian*

# Remarks on Blochfunctions

1. Quantum number  $k$ : Crystal momentum

$$\frac{\hbar}{i} \nabla \Psi_{\mathbf{k}} = \frac{\hbar}{i} \nabla e^{i \mathbf{k} \cdot \mathbf{r}} u_{\mathbf{k}}(\vec{\mathbf{r}}) = \hbar \mathbf{k} \cdot \Psi_{\mathbf{k}} + \frac{\hbar}{i} e^{i \mathbf{k} \cdot \mathbf{r}} \nabla u_{\mathbf{k}}(\vec{\mathbf{r}})$$

2.  $k$  may be confined to the 1<sup>st</sup> BZ (Reduced BZ)

$$\Psi_{\vec{\mathbf{k}}+\vec{\mathbf{G}}} = e^{i \vec{\mathbf{k}} \cdot \vec{\mathbf{r}}} u_{\vec{\mathbf{k}}+\vec{\mathbf{G}}}(\vec{\mathbf{r}})$$

3. Fourier expansion Bloch functions

Periodicity  $\Psi(\vec{\mathbf{r}} + p_i \cdot \vec{\mathbf{a}}_i) = \Psi(\vec{\mathbf{r}})$

$$u_{\mathbf{k}}(\vec{\mathbf{r}}) = \sum_{\vec{\mathbf{G}}} C_{\vec{\mathbf{k}}+\vec{\mathbf{G}}} \cdot e^{i \vec{\mathbf{G}} \cdot \vec{\mathbf{r}}}$$

4. Band structure  $\varepsilon_n(\vec{\mathbf{k}})$

# Band structure: Approaches

## Last time

- Empty lattice (only periodicity)
- Perturbation theory (nearly free electrons, weak potential)

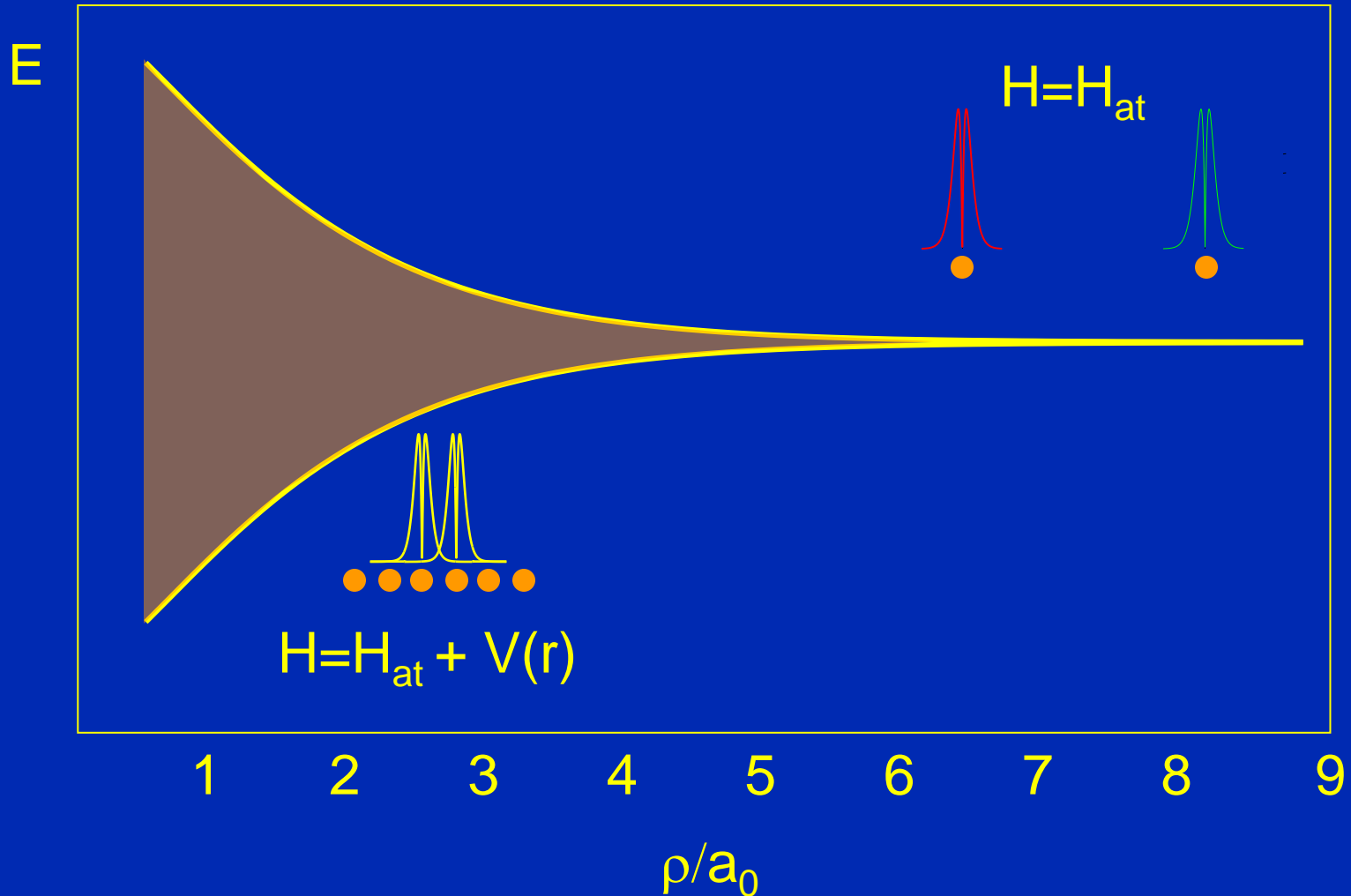
## Today

- Tight binding method (LCAO)

## Other methods

- Exact models (Kronig-Penney model, see book)
- ‘advanced’ methods: Ashcroft & Mermin chapter 11

# Tight binding approach



# Tight binding

Atomic orbitals  $\varphi(\mathbf{r})$  :  $H_{\text{at}} \varphi(\mathbf{r}) = E_{\text{at}} \varphi(\mathbf{r})$

Solid state:  $H = H_{\text{at}} + V(\mathbf{r})$

$$\Rightarrow \Psi_{\vec{k}}(\vec{r}) = \frac{1}{\sqrt{N}} \sum_j e^{i\vec{k} \cdot \vec{r}_j} \cdot \varphi(\vec{r} - \vec{r}_j)$$

1<sup>st</sup> order correction to energy: diagonal elements

$$\begin{aligned} \langle \mathbf{k} | H | \mathbf{k} \rangle &= \frac{1}{N} \sum_j \sum_m e^{i\vec{k} \cdot (\vec{r}_j - \vec{r}_m)} \cdot \langle \varphi_m | H | \varphi_j \rangle \\ &= \sum_m e^{i\vec{k} \cdot \vec{\rho}_m} \cdot \int dV \varphi^*(\vec{r} - \vec{\rho}_m) H \varphi(\vec{r}) \end{aligned}$$

# Tight binding

$$\langle \mathbf{k} | \mathbf{H} | \mathbf{k} \rangle = \sum_{\mathbf{m}} e^{i \vec{k} \cdot \vec{\rho}_{\mathbf{m}}} \cdot \int dV \varphi^*(\vec{r} - \vec{\rho}_{\mathbf{m}}) \mathbf{H} \varphi(\vec{r})$$

Only nearest neighbours:

Self  $\rho = 0 \Rightarrow \int dV \varphi^*(\vec{r}) \mathbf{H} \varphi(\vec{r}) \equiv -\alpha$

N.N.  $\rho_{\mathbf{m}} = \rho \Rightarrow \int dV \varphi^*(\vec{r} - \vec{\rho}) \mathbf{H} \varphi(\vec{r}) \equiv -\gamma$



$$\varepsilon(\mathbf{k}) = \langle \mathbf{k} | \mathbf{H} | \mathbf{k} \rangle = -\alpha - \gamma \cdot \sum_{\mathbf{nn}} e^{-i \vec{k} \cdot \vec{\rho}_{\mathbf{nn}}}$$

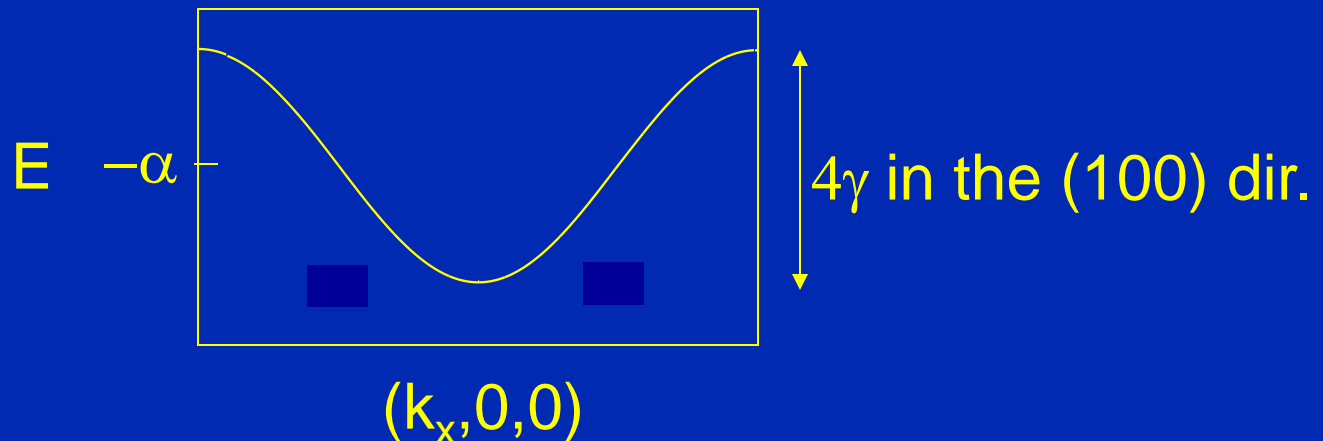
# Tight binding

$$\varepsilon(\mathbf{k}) = \langle \mathbf{k} | \mathbf{H} | \mathbf{k} \rangle = -\alpha - \gamma \cdot \sum_{\mathbf{nn}} e^{-i \vec{k} \cdot \vec{\rho}_{\mathbf{nn}}}$$

S.C.: n.n. at  $(\pm a, 0, 0)$ ;  $(0, \pm a, 0)$ ;  $(0, 0, \pm a)$



$$\varepsilon(\mathbf{k}) = -\alpha - 2\gamma [ \cos(k_x a) + \cos(k_y a) + \cos(k_z a) ]$$



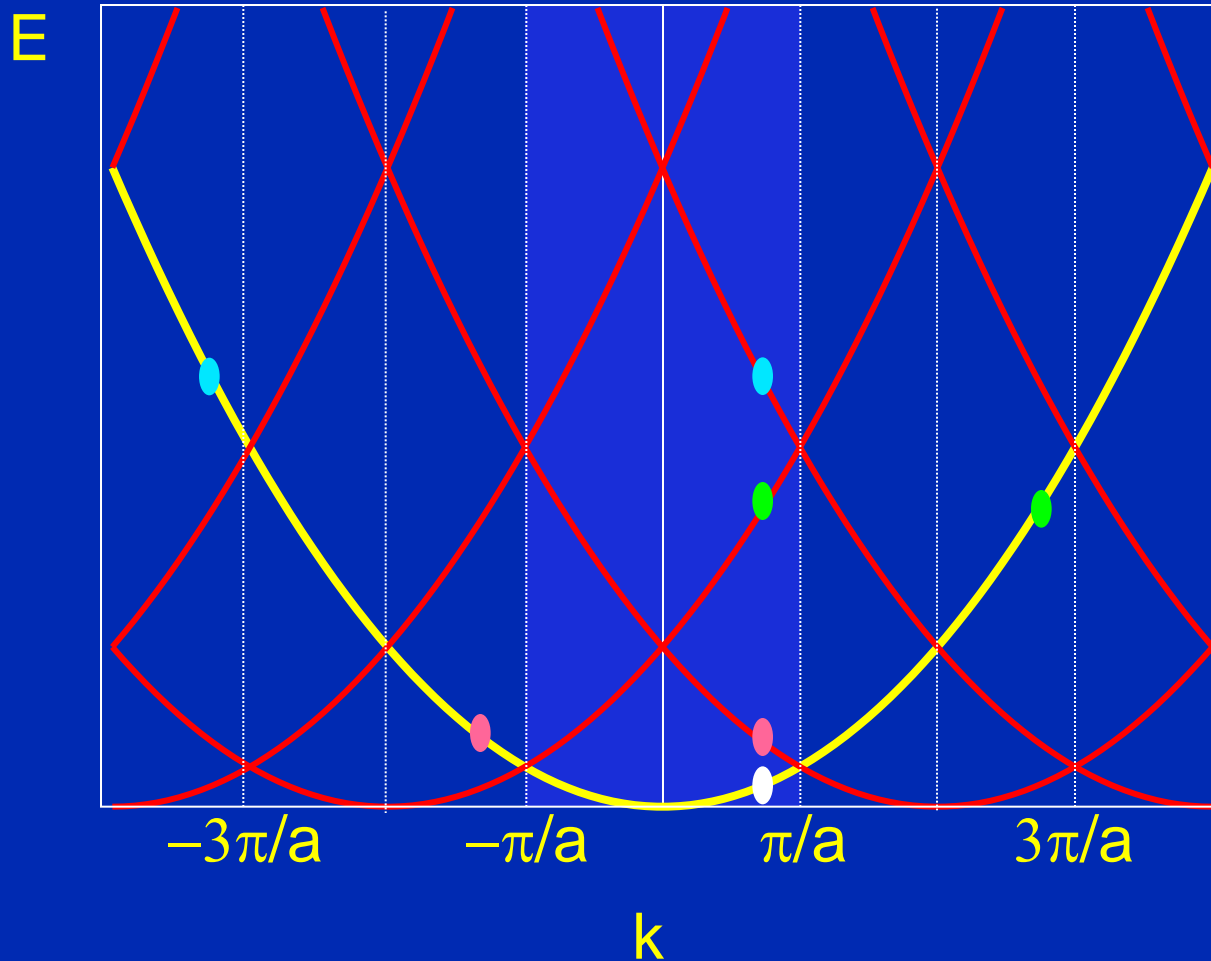


# Tight binding

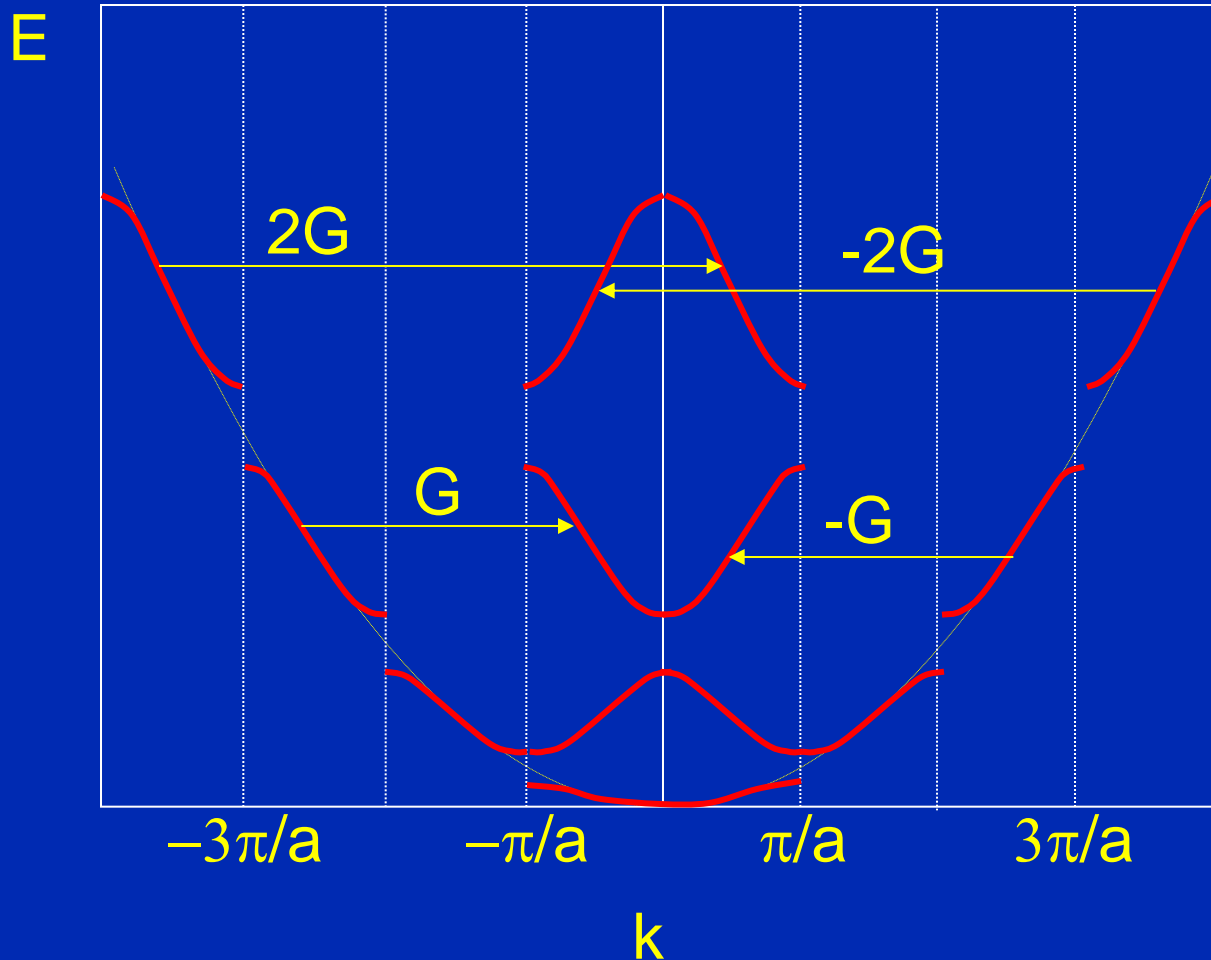
- TB method: Insulators and core electrons and d-electrons in transition metals (e.g. Cu, V, Ti etc.).
- Nearly free electron model: conduction electrons, s or p, alkali (e.g. Na, K, etc.)

period	group 1*	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
	Ia	IIa	IIIb	IVb	Vb	VIb	VIIb	VIIIb	VIIIb	VIIIb	Ib	IIb	IIIa	IVa	Va	VIa	VIIa	VIIIb	0
1	H	He																	
2	Li	Be											B	C	N	O	F	Ne	
3	Na	Mg	Al	Si	P	S	Cl	Ar					Al	Si	P	S	Cl	Ar	
4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
6	Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	
7	Fr	Ra	Ac	****	****	****	****	****	****	****	****	****							
6	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu					
7	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr					

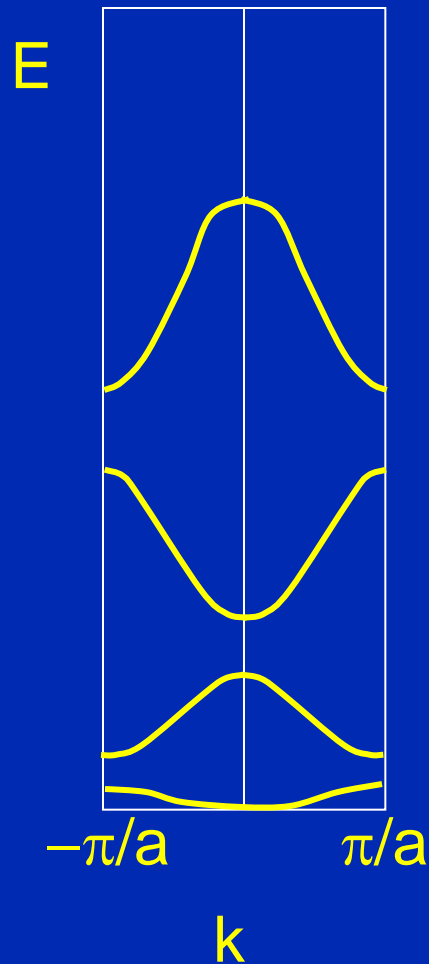
# Reduced Brillouin zone



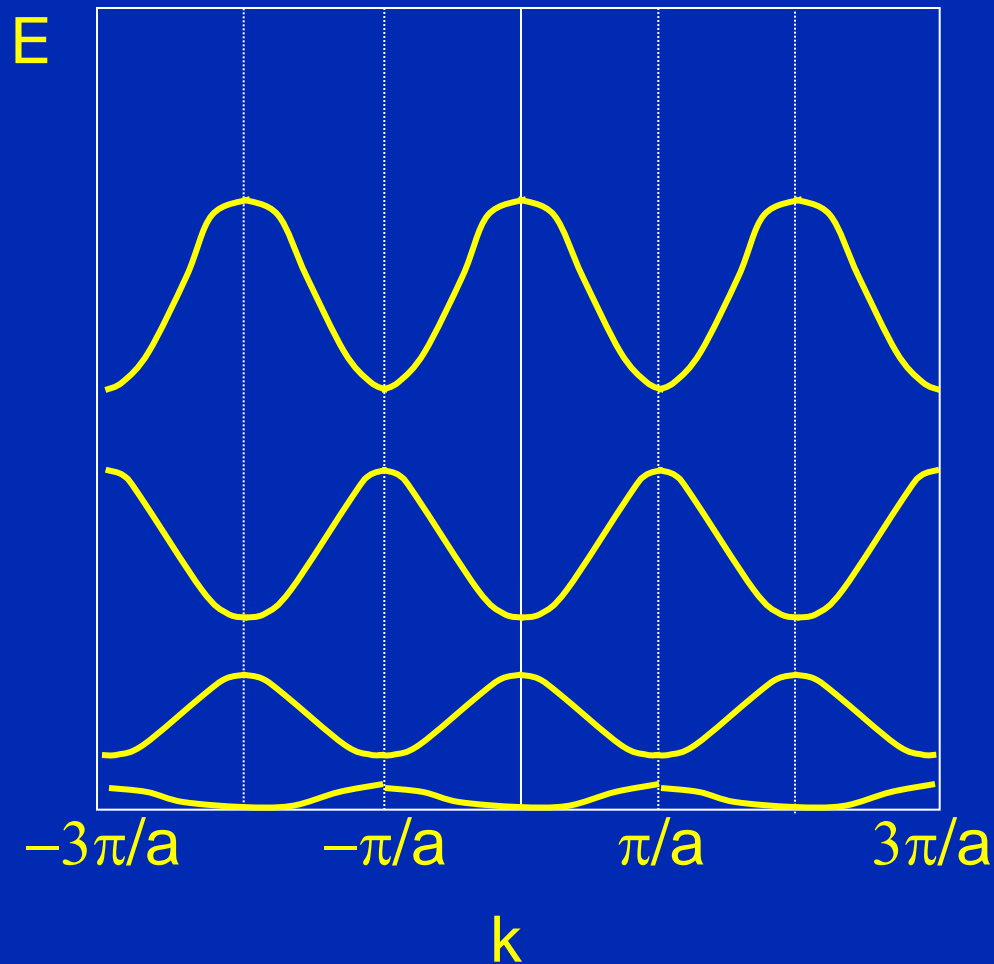
# Extended zone scheme



# Reduced zone scheme



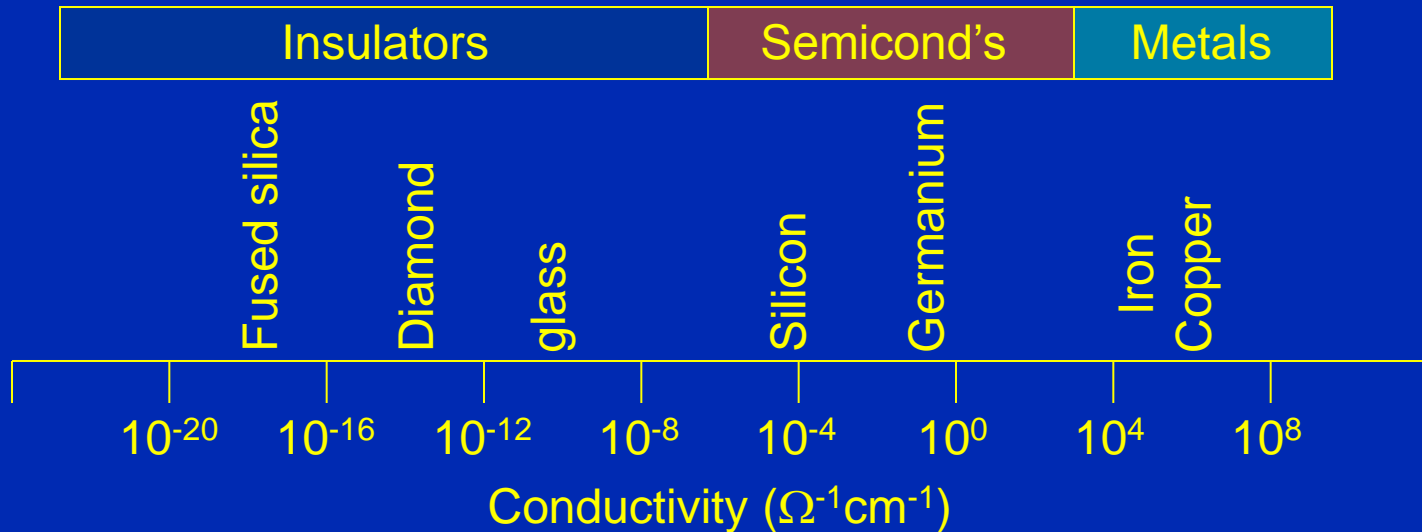
# Periodic zone scheme



$$E_{n, \vec{k} + \vec{G}} \equiv E_{n, \vec{k}}$$

# Bandstructure

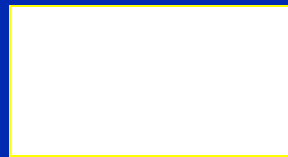
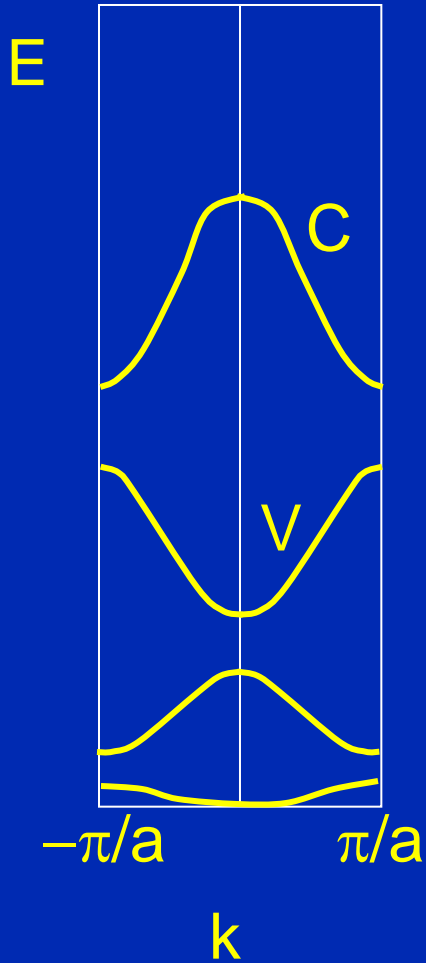
- Why are not all solids metals



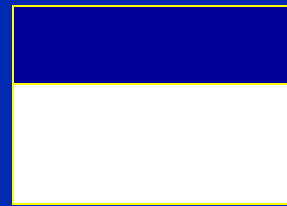
- Why is diamond an insulator
- Why is copper a metal
- Why does the conductivity of Si increase upon heating

# Conductivity

$$\sigma = n \cdot e \cdot \mu$$



Insulator



Metal



p-Doped  
Semi-  
Conductor



n-Doped  
Semi-  
Conducto

# Diamond and Germanium

Germanium ( $E_g=0.7$  eV)

Diamond ( $E_g=5.5$  eV)

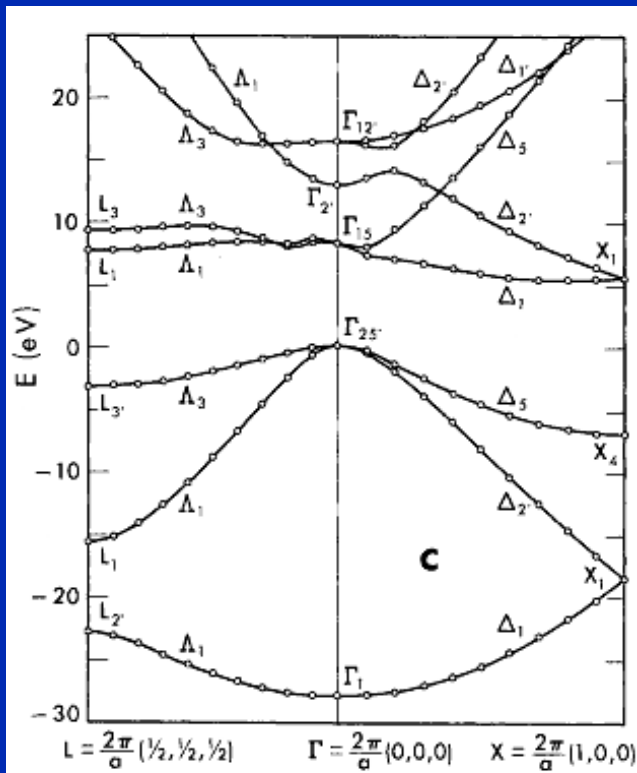
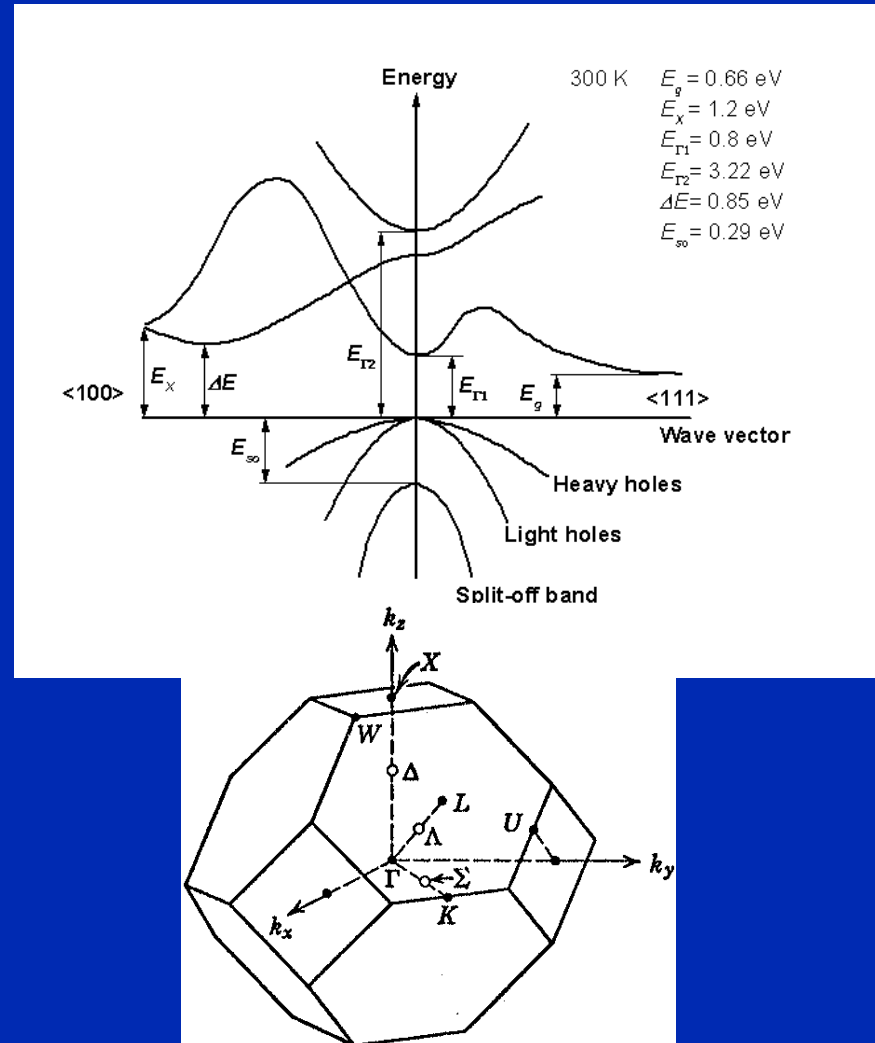
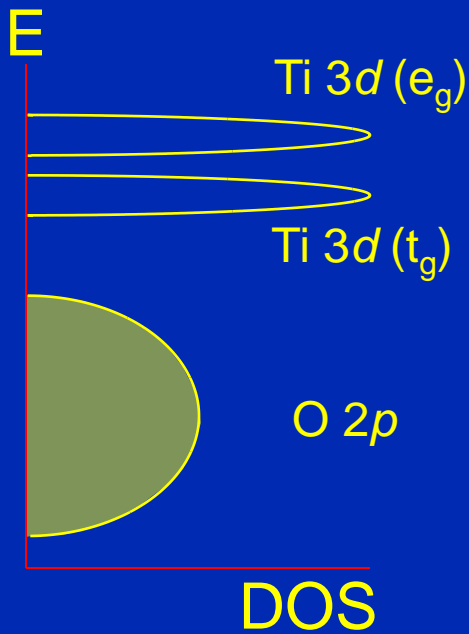


FIG. 1. The electronic band structure of diamond.

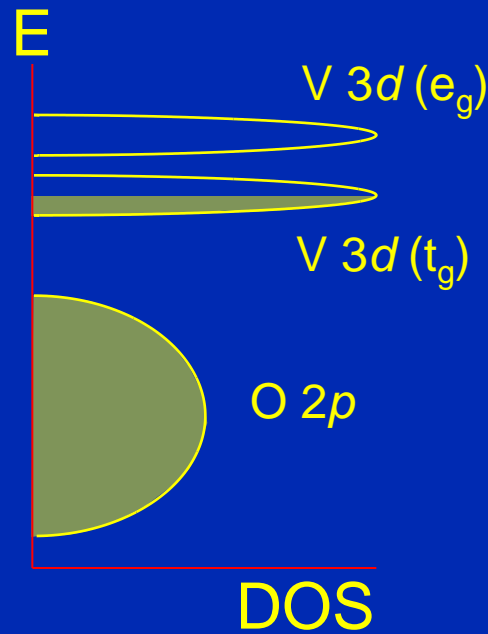




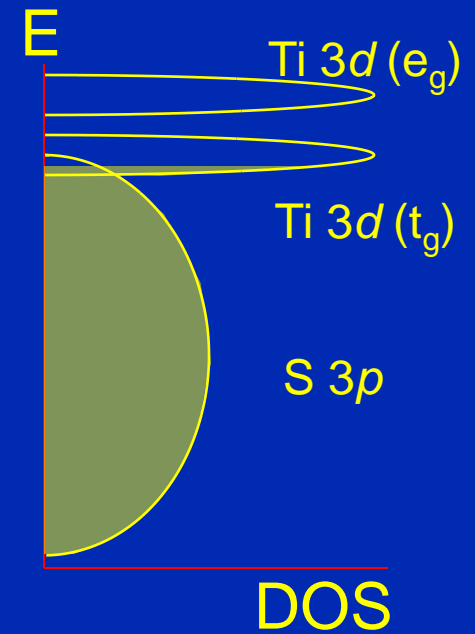
# Bandfilling & overlap



$\text{TiO}_2$   
 Empty  $d$ -bands



$\text{VO}_2$   
 Partially filled  $d$ -bands



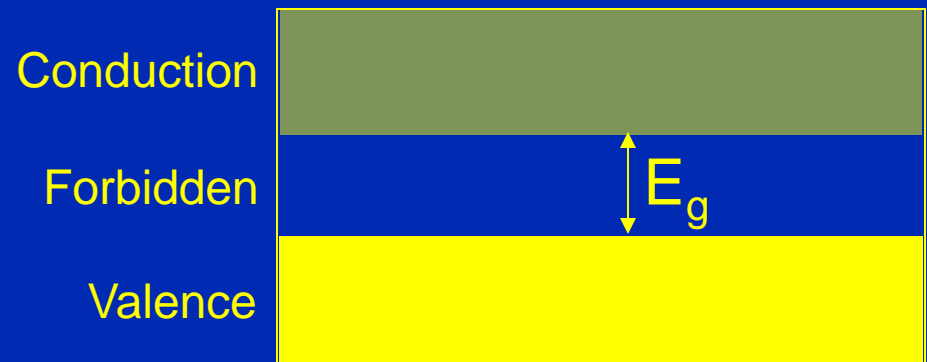
$\text{TiS}_2$   
 Overlapping bands

Ti:  $[\text{Ar}]3d^24s^2$

V:  $[\text{Ar}]3d^34s^2$

# Band gap in some materials

		$E_g$ (eV)
Diamond	I	5.4
Si	I	1.17
Ge	I	0.74
GaAs	d	1.52
GaSb	d	0.81
Te	d	0.33
ZnO		3.44
Cu <sub>2</sub> O	d	2.17
CdTe	d	1.6
TiO <sub>2</sub>		3.03



➔ Direct & indirect  
Optical absorption

