

# Condensed Matter Physics II

## SS 2015

Wednesday 9:30-12:30

Seminar Room Physics 2

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## OXIDES WHICH SHOW A METAL-TO-INSULATOR TRANSITION AT THE NEEL TEMPERATURE

F. J. Morin

Bell Telephone Laboratories, Murray Hill, New Jersey

(Received June 5, 1959)

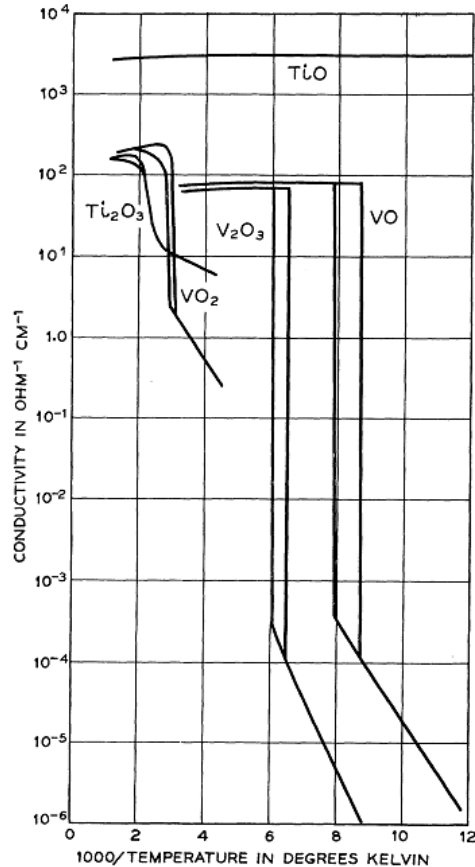


FIG. 1. Conductivity as a function of reciprocal temperature for the lower oxides of titanium and vanadium. Measurements were made along the [100] direction in VO, and along the *c* axis in  $V_2O_3$  and  $VO_2$ .



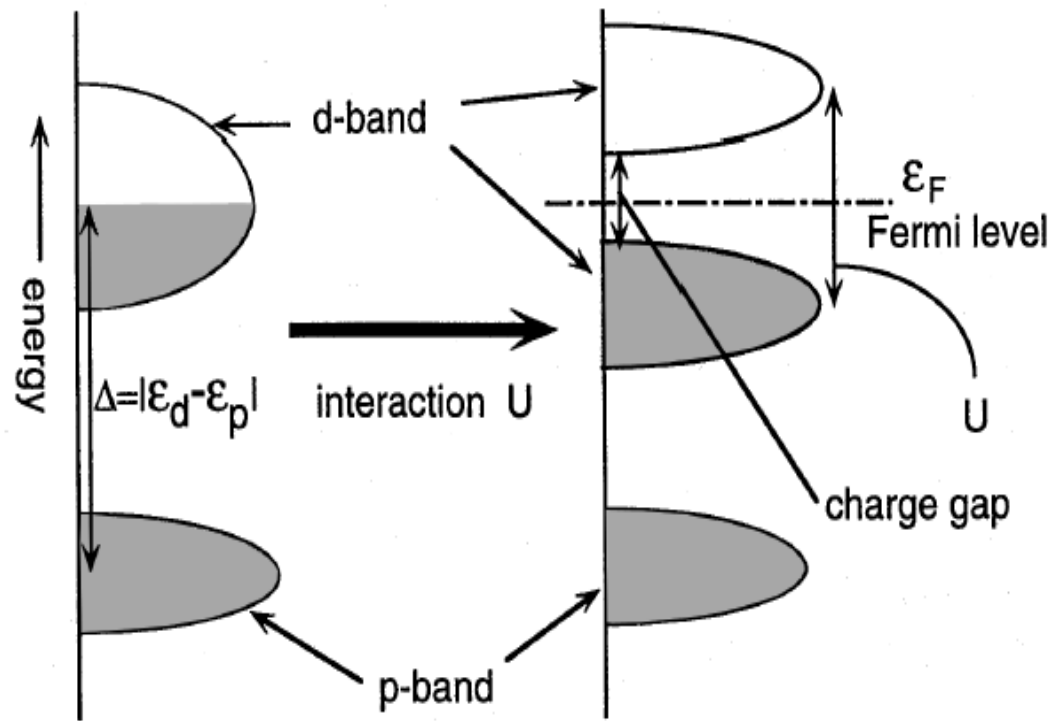
Metal-Insulator transition at 340 K,  
also structural change: low T monoclinic,  
high T tetragonal (rutile) !!

Band insulator?

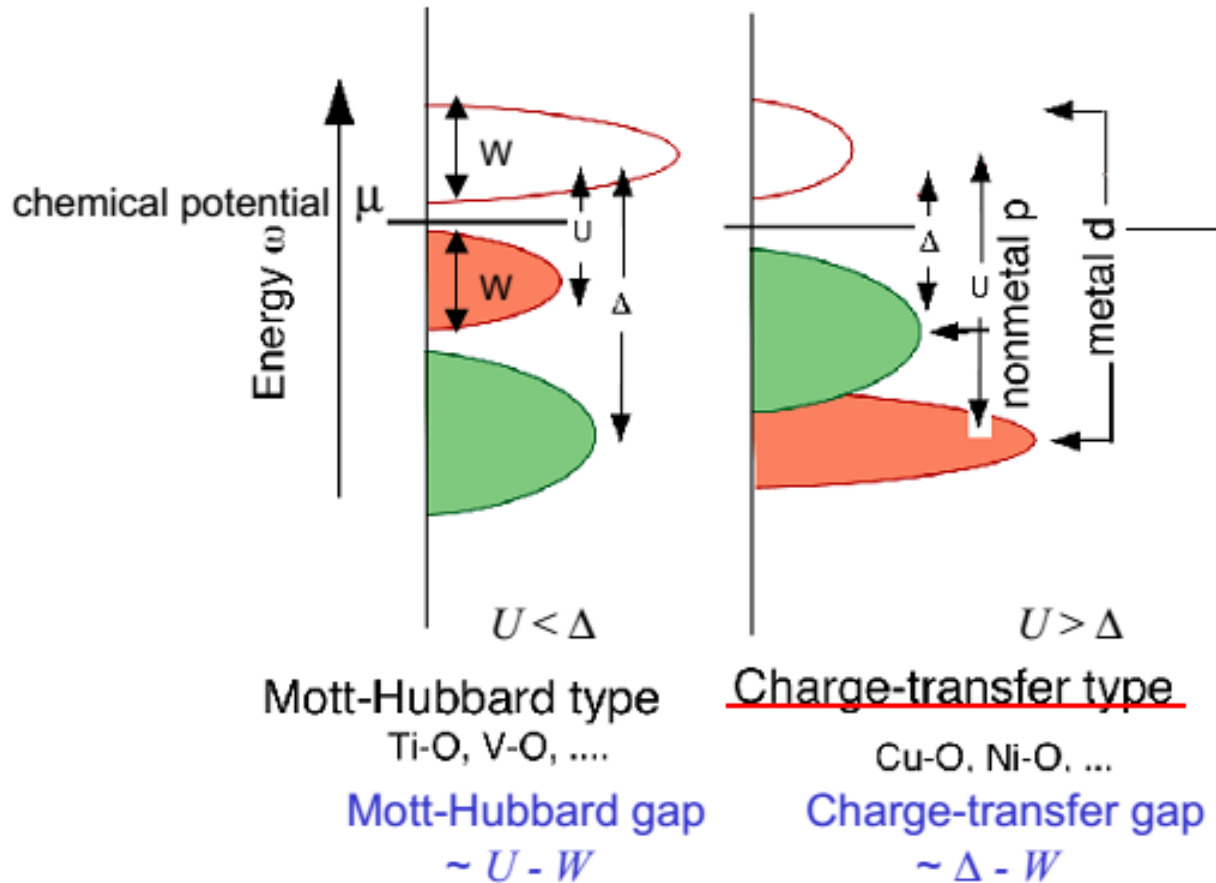
Mott Hubbard insulator?

Valence electrons V:  $3d^34s^2$

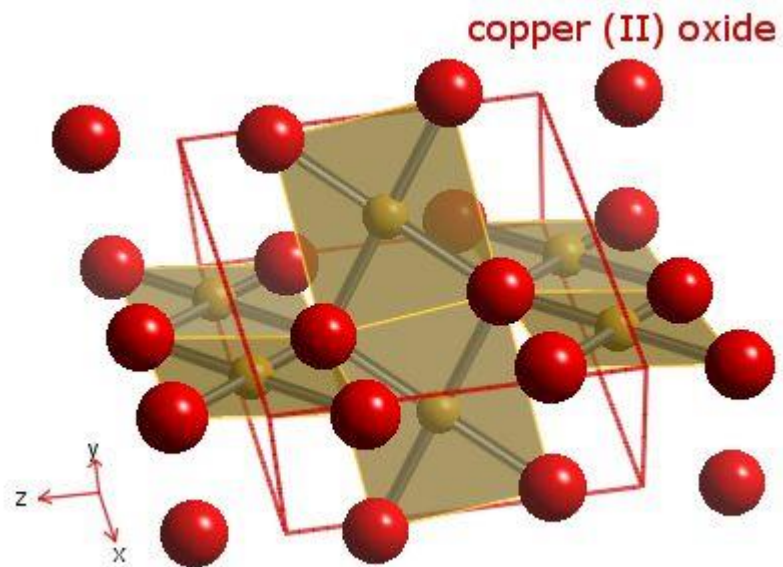
So  $V^{4+} : 3d^1$



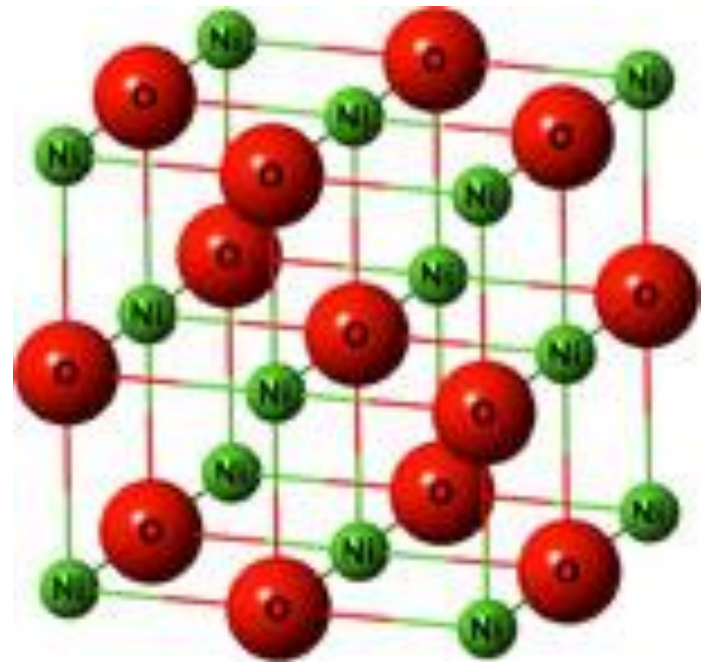
# Mott vs Charge transfer



# CuO & NiO

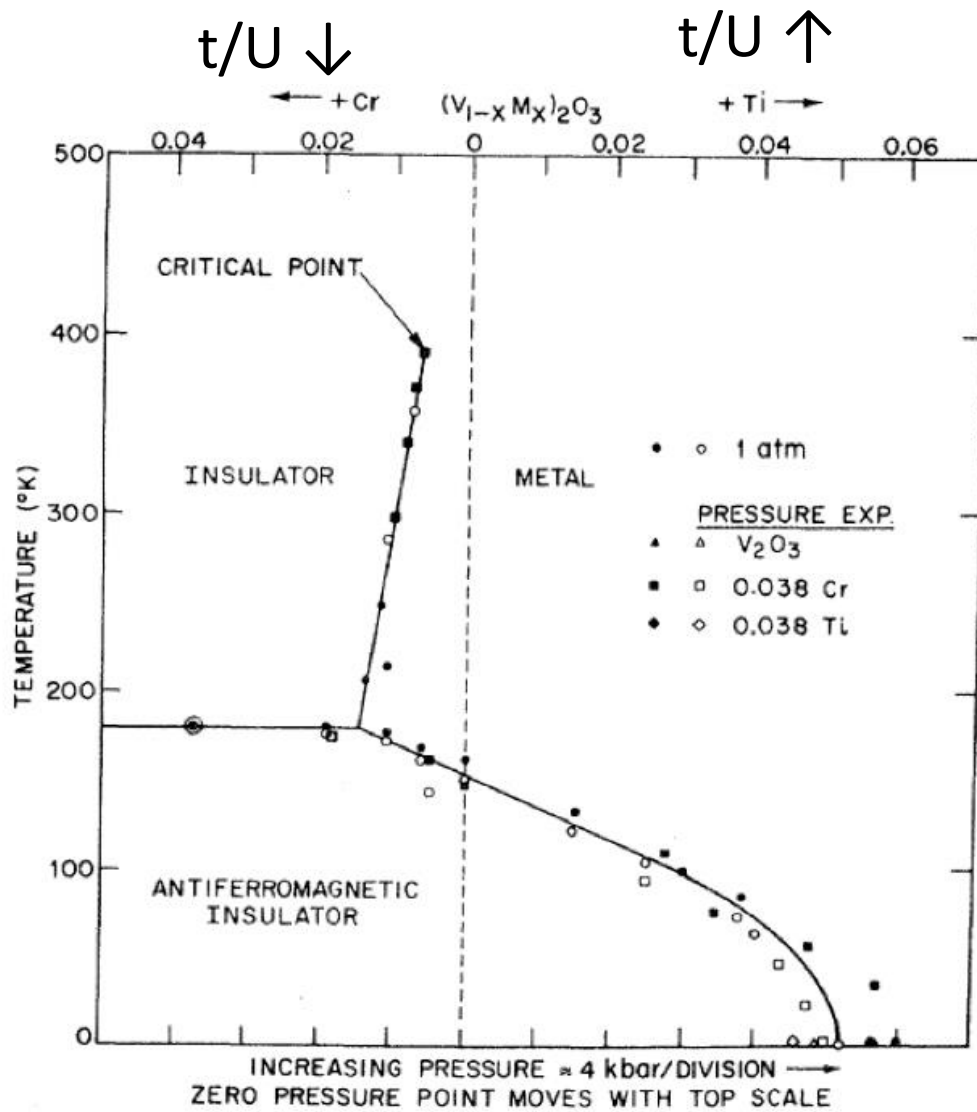


CuO



NiO





$U \uparrow$  →

$U \downarrow$  ↓

Sc <sup>21</sup>	Ti <sup>22</sup>	V <sup>23</sup>	Cr <sup>24</sup>	Mn <sup>25</sup>	Fe <sup>26</sup>	Co <sup>27</sup>	Ni <sup>28</sup>	Cu <sup>29</sup>	Zn <sup>30</sup>
3d 4s <sup>2</sup>	3d <sup>2</sup> 4s <sup>2</sup>	3d <sup>3</sup> 4s <sup>2</sup>	3d <sup>5</sup> 4s	3d <sup>5</sup> 4s <sup>2</sup>	3d <sup>6</sup> 4s <sup>2</sup>	3d <sup>7</sup> 4s <sup>2</sup>	3d <sup>8</sup> 4s <sup>2</sup>	3d <sup>10</sup> 4s	3d <sup>10</sup> 4s <sup>2</sup>
Y <sup>39</sup>	Zr <sup>40</sup>	Nb <sup>41</sup>	Mo <sup>42</sup>	Tc <sup>43</sup>	Ru <sup>44</sup>	Rh <sup>45</sup>	Pd <sup>46</sup>	Ag <sup>47</sup>	Cd <sup>48</sup>
4d 5s <sup>2</sup>	4d <sup>2</sup> 5s <sup>2</sup>	4d <sup>4</sup> 5s	4d <sup>5</sup> 5s	4d <sup>6</sup> 5s	4d <sup>7</sup> 5s	4d <sup>8</sup> 5s	4d <sup>10</sup> -	4d <sup>10</sup> 5s	4d <sup>10</sup> 5s <sup>2</sup>
La <sup>57</sup>	Hf <sup>72</sup>	Ta <sup>73</sup>	W <sup>74</sup>	Re <sup>75</sup>	Os <sup>76</sup>	Ir <sup>77</sup>	Pt <sup>78</sup>	Au <sup>79</sup>	Hg <sup>80</sup>
5d 6s <sup>2</sup>	5d <sup>2</sup> 6s <sup>2</sup>	5d <sup>3</sup> 6s <sup>2</sup>	5d <sup>4</sup> 6s <sup>2</sup>	5d <sup>5</sup> 6s <sup>2</sup>	5d <sup>6</sup> 6s <sup>2</sup>	5d <sup>9</sup> -	5d <sup>9</sup> 6s	5d <sup>10</sup> 6s	5d <sup>10</sup> 6s <sup>2</sup>
Ac <sup>89</sup>									
6d 7s <sup>2</sup>	Ce <sup>58</sup>	Pr <sup>59</sup>	Nd <sup>60</sup>	Pm <sup>61</sup>	Sm <sup>62</sup>	Eu <sup>63</sup>	Gd <sup>64</sup>	Tb <sup>65</sup>	Dy <sup>66</sup>
	4f <sup>2</sup>	4f <sup>3</sup>	4f <sup>4</sup>	4f <sup>5</sup>	4f <sup>6</sup>	4f <sup>7</sup>	4f <sup>7</sup>	4f <sup>8</sup>	4f <sup>9</sup>
	6s <sup>2</sup>	6s <sup>2</sup>	6s <sup>2</sup>	6s <sup>2</sup>	6s <sup>2</sup>	6s <sup>2</sup>	6s <sup>2</sup>	6s <sup>2</sup>	6s <sup>2</sup>
	Th <sup>90</sup>	Pa <sup>91</sup>	U <sup>92</sup>	Np <sup>93</sup>	Pu <sup>94</sup>	Am <sup>95</sup>	Cm <sup>96</sup>	Bk <sup>97</sup>	Cf <sup>98</sup>
	-	5f <sup>2</sup>	5f <sup>3</sup>	5f <sup>3</sup>	5f <sup>6</sup>	5f <sup>7</sup>	5f <sup>7</sup>	5f <sup>7</sup>	5f <sup>8</sup>
	6d <sup>2</sup>	6d	6d	7s <sup>2</sup>	7s <sup>2</sup>	7s <sup>2</sup>	7s <sup>2</sup>	6d	6d
	7s <sup>2</sup>	7s <sup>2</sup>	7s <sup>2</sup>	7s <sup>2</sup>	7s <sup>2</sup>	7s <sup>2</sup>	7s <sup>2</sup>	7s <sup>2</sup>	7s <sup>2</sup>

# Evidence for a structurally-driven insulator-to-metal transition in VO<sub>2</sub>: A view from the ultrafast timescale

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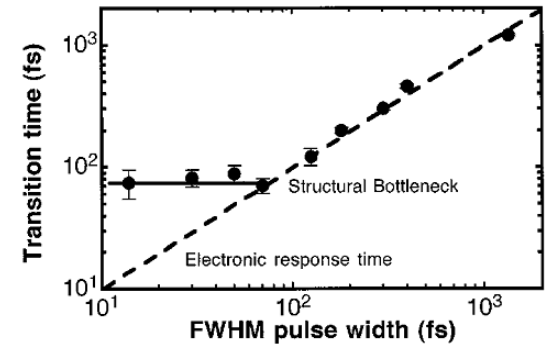
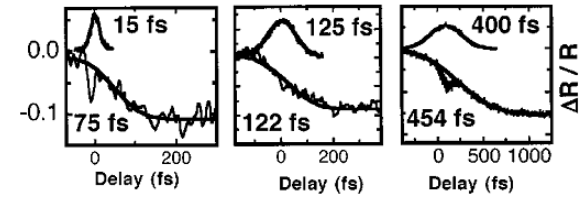
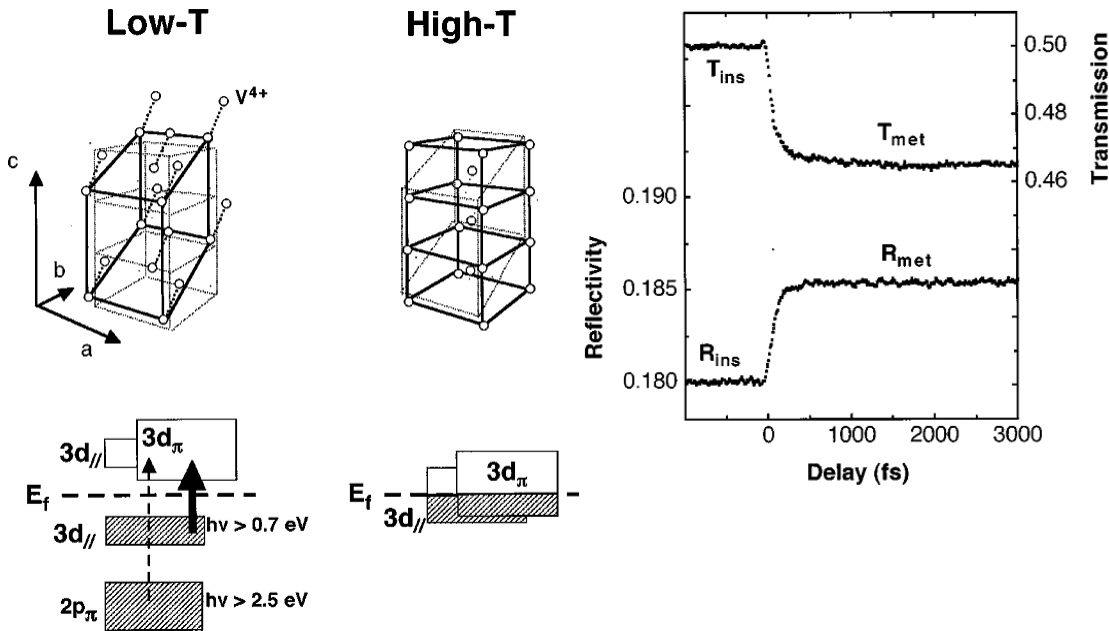
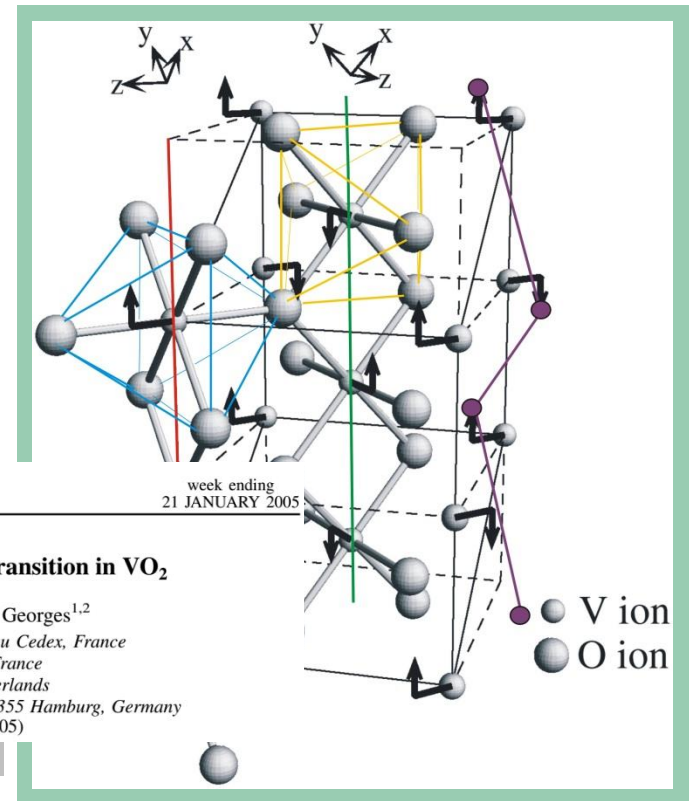
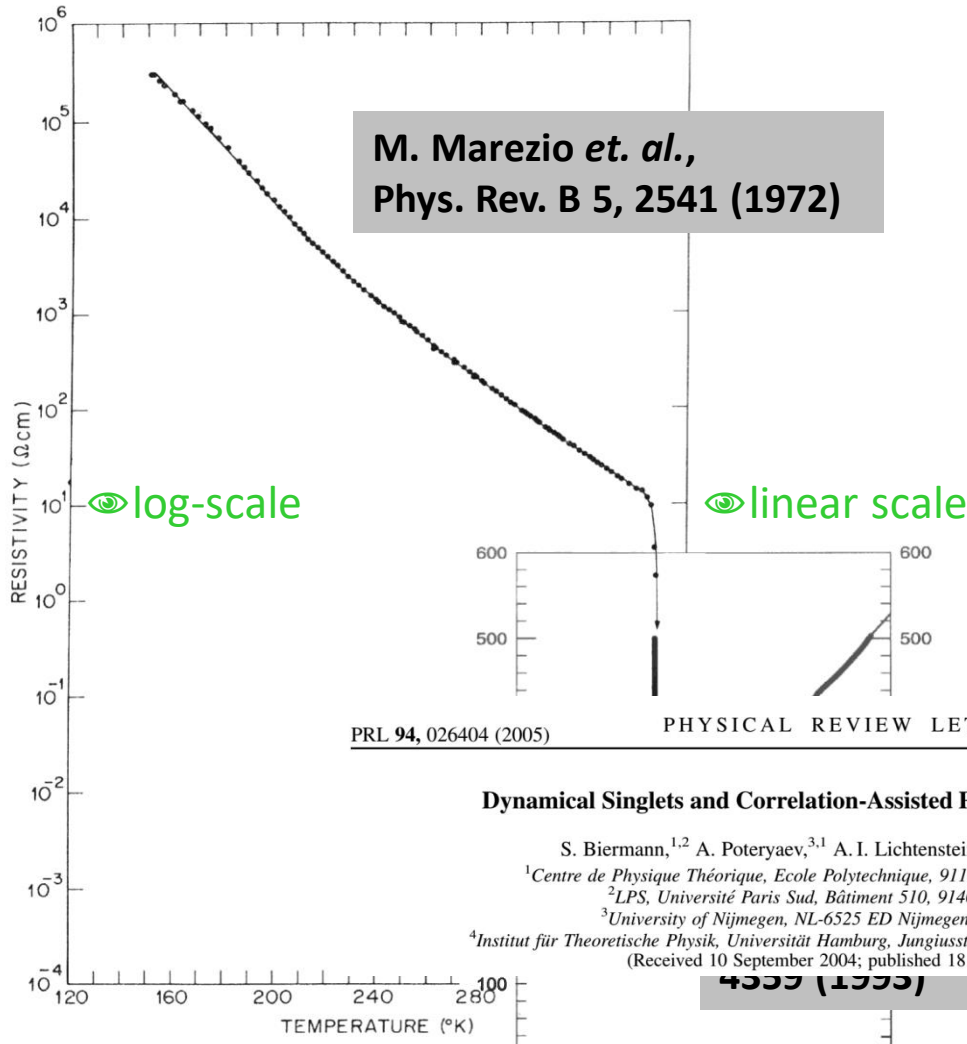


FIG. 3. Pump-probe reflectivity experiments of the photoinduced phase transition. The experiments are performed in the non-etched structure with variable pulse durations between 1.5 ps and 15 fs, as measured at the sample position. White light was amplified in 1-mm, 32°-cut BBO, pumped with 400-nm pulses crossing the seed light at 3.5°. Pulse compression in a pair of prisms was used to minimize the duration of the pump-probe autocorrelation at the sample position. The experiments were conducted using pulses of 100-nm bandwidth full width at half maximum centered around 650 nm.

# Metal insulator transition in VO<sub>2</sub> at 340 K

**$T > 340\text{K}$ : metal, rutile**

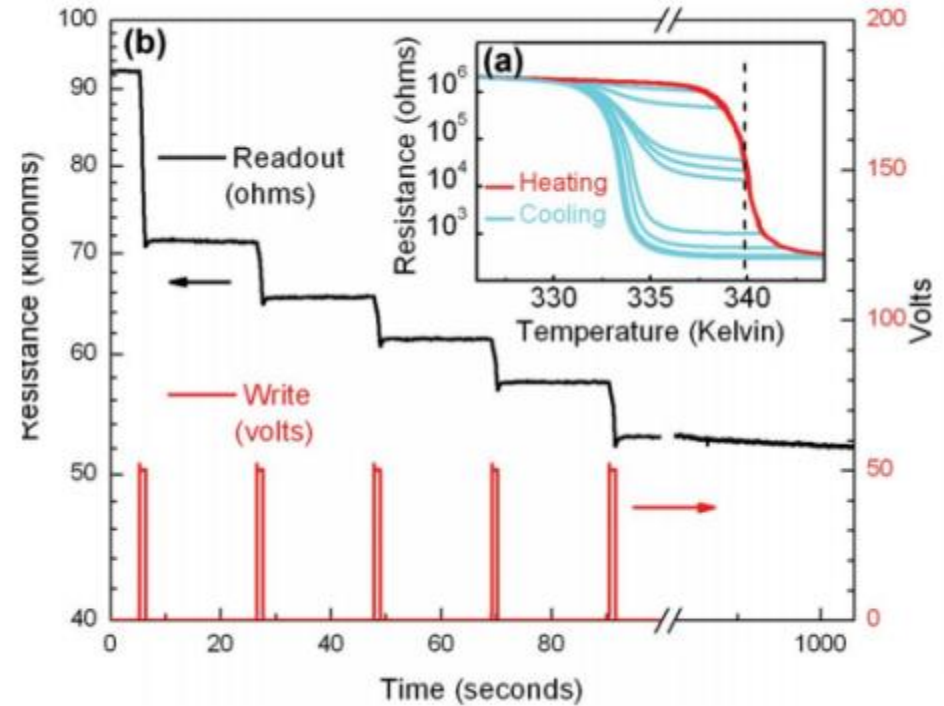
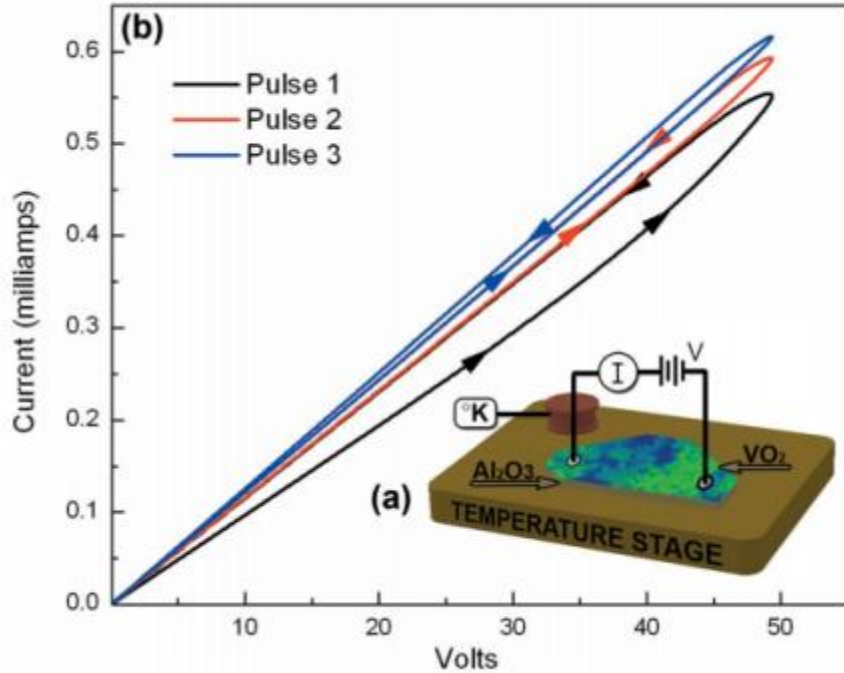
**$T < 340\text{K}$ : insulator, monoclinic, dimerized zig-zag chain**



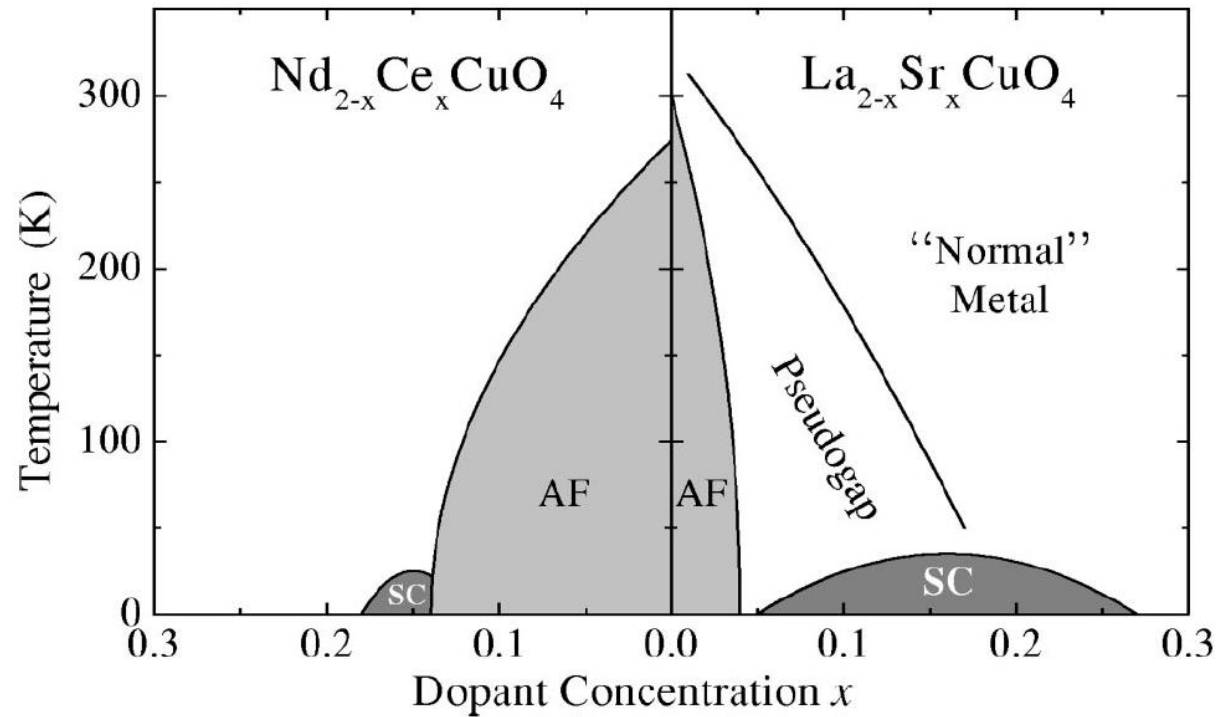
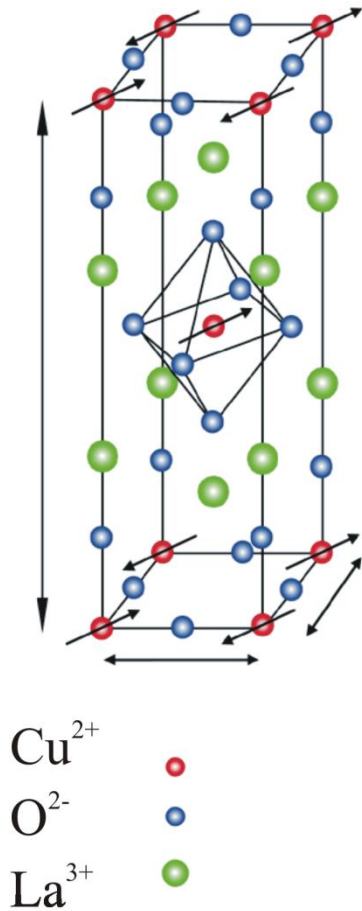
Dimerization → Peierls transition = band structure effect  
 Combination of Peierls and Hubbard physics



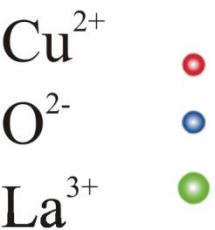
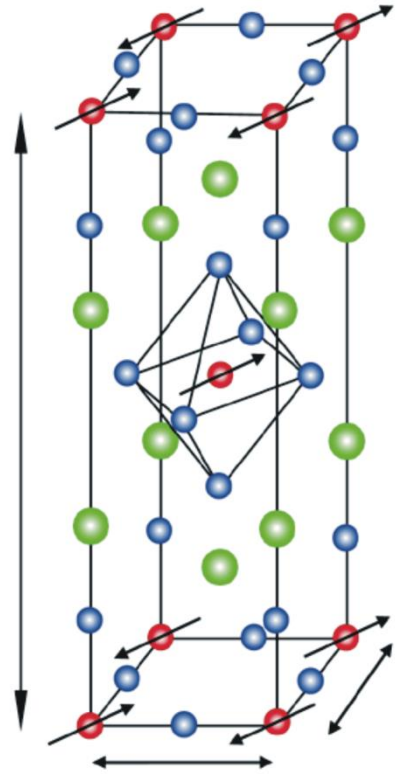
# VO<sub>2</sub> Memristors



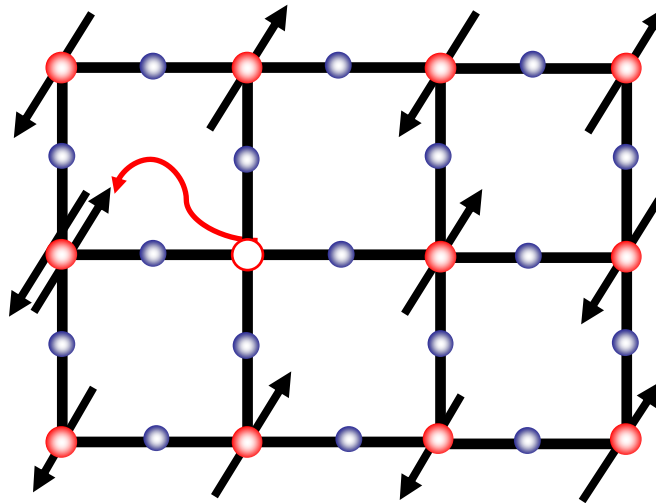
# High temperature superconductivity



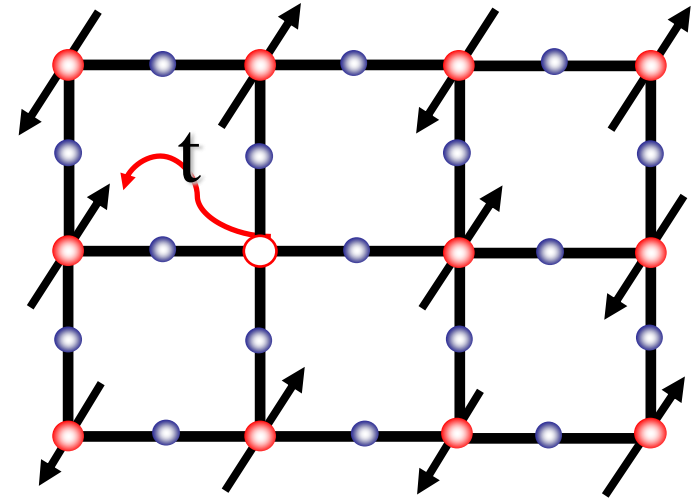
# CuO plane: strongly-correlated electron system



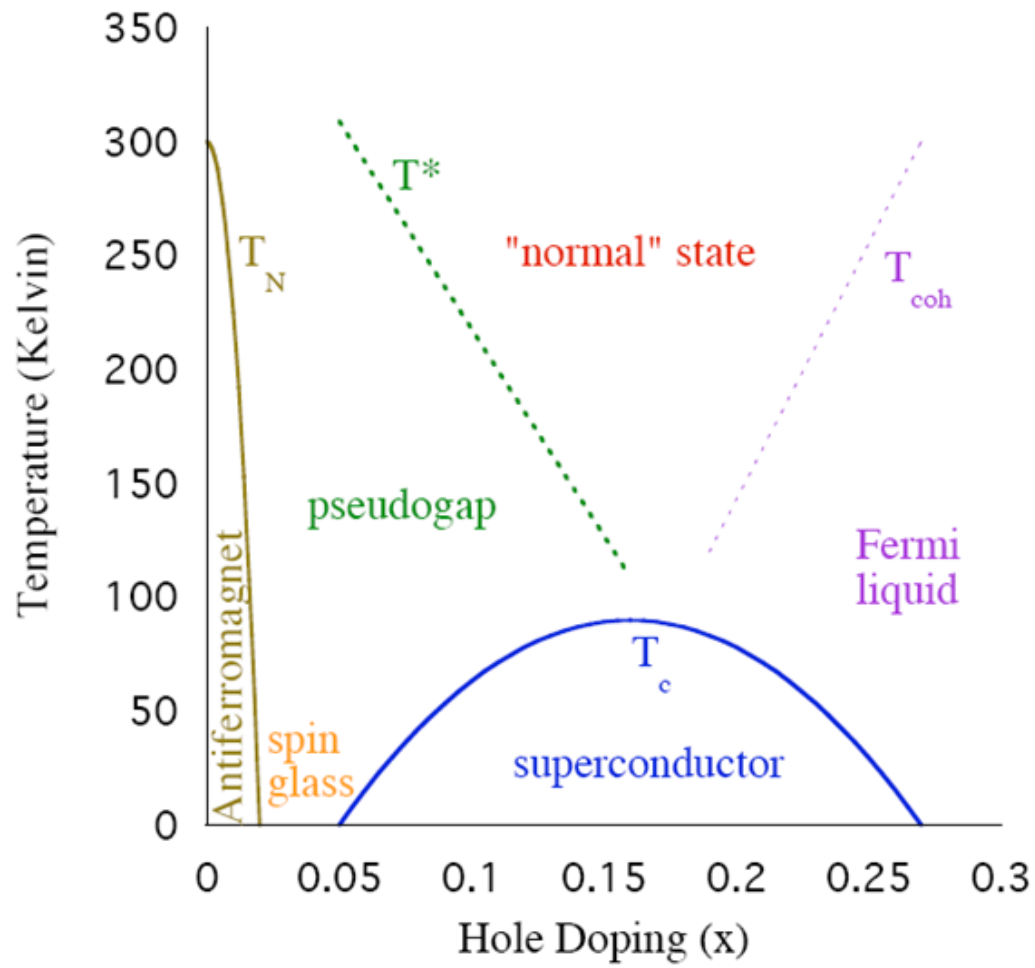
One hole per site: should be a metal according to band theory.  
**Mott insulator.**



*Undoped CuO<sub>2</sub> plane:  
Mott Insulator due to  
 $e^- - e^-$  interaction  
Virtual hopping induces  
AF exchange  $J=4t^2/U$*

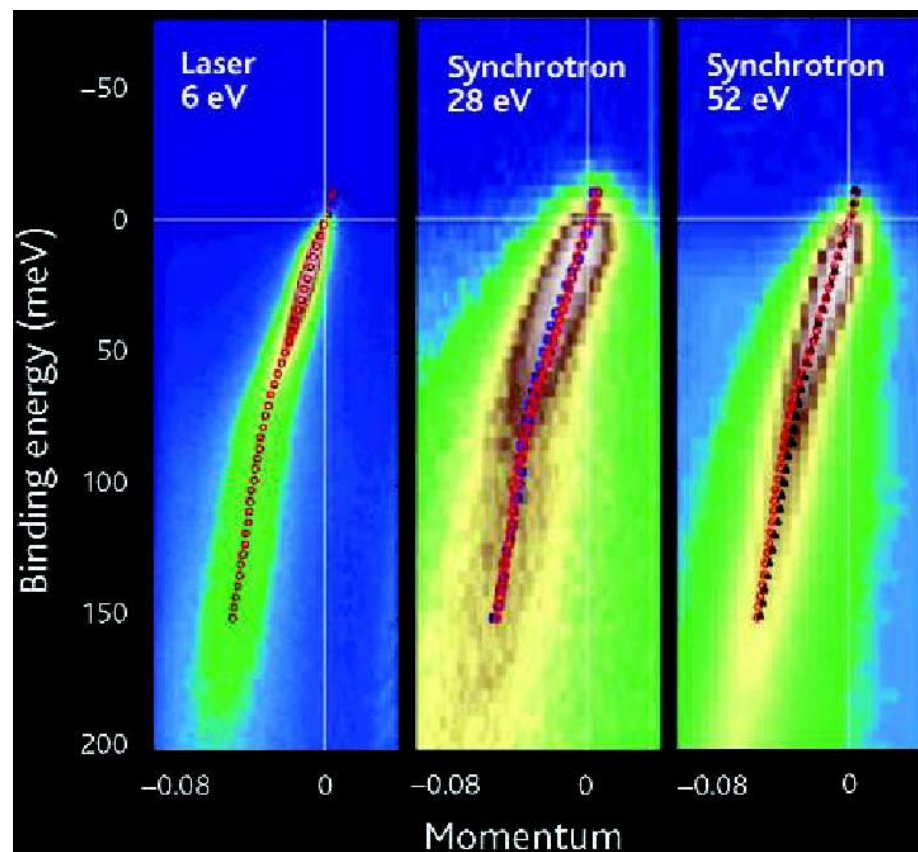


*CuO<sub>2</sub> plane with doped holes:  
 $\text{La}^{3+} \rightarrow \text{Sr}^{2+}$ :  $\text{La}_{2-x}\text{Sr}_x\text{CuO}_4$*

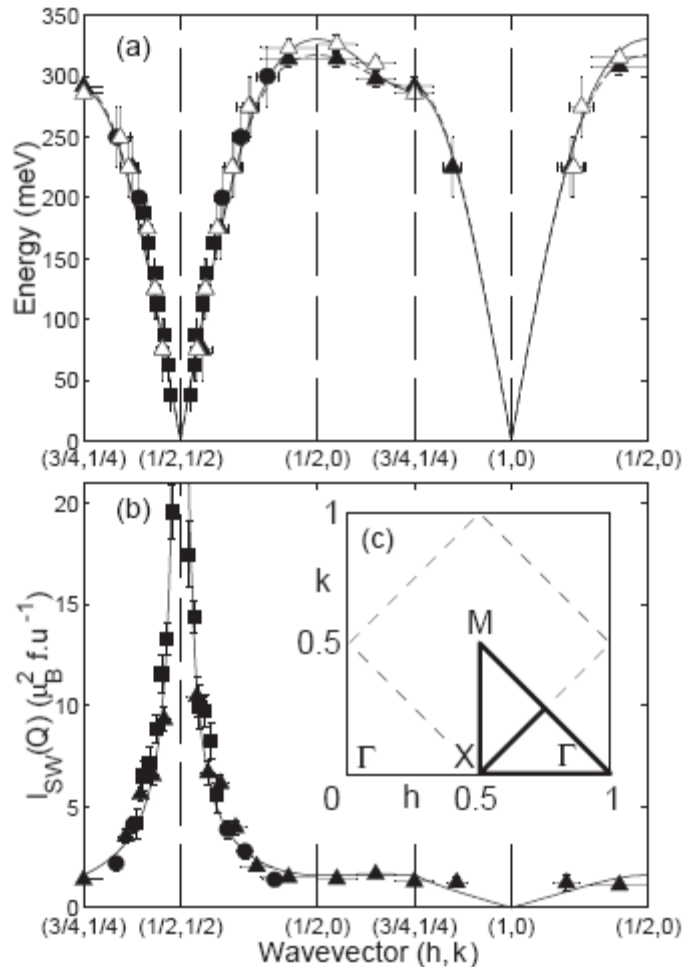


## Laser Based Angle-Resolved Photoemission, the Sudden Approximation, and Quasiparticle-Like Spectral Peaks in $\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_{8+\delta}$

J. D. Koralek,<sup>1,2,\*</sup> J. F. Douglas,<sup>1</sup> N. C. Plumb,<sup>1</sup> Z. Sun,<sup>1,3</sup> A. V. Fedorov,<sup>3</sup> M. M. Murnane,<sup>1,2</sup> H. C. Kapteyn,<sup>1,2</sup>  
S. T. Cundiff,<sup>2</sup> Y. Aiura,<sup>4</sup> K. Oka,<sup>4</sup> H. Eisaki,<sup>4</sup> and D. S. Dessau<sup>1,2,†</sup>



Science **310**, 1271 (2005)



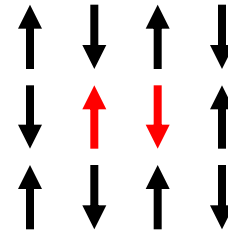
$$J = 146(4) \text{ meV}$$

By fitting the spin wave dispersion measured by neutron scattering. (also needs a small ring exchange term.)

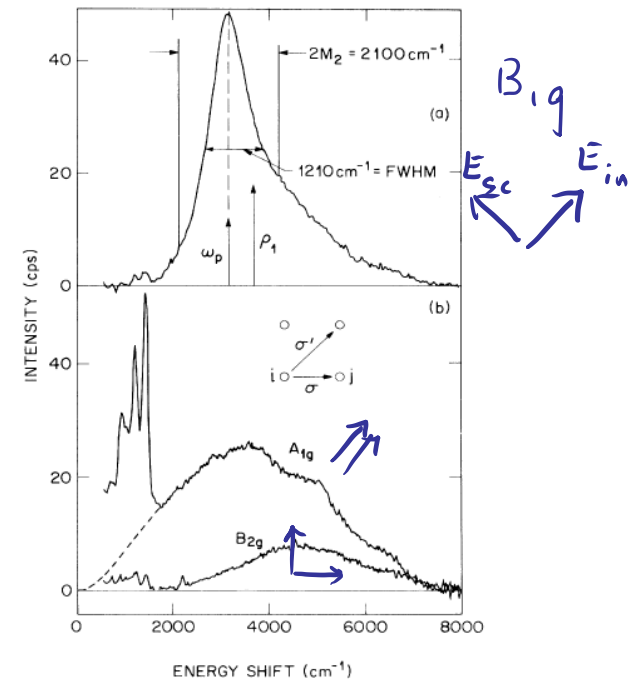
Also from Raman scattering.

$$I(\omega) = \sum_i \delta(\omega - (E_i - E_0)) |\langle 0 | H_R | i \rangle|^2$$

$$H_R = \sum_{\langle ij \rangle} (\mathbf{E}_{inc} \cdot \boldsymbol{\sigma}_{ij}) (\mathbf{E}_{sc} \cdot \boldsymbol{\sigma}_{ij}) \mathbf{S}_i \cdot \mathbf{S}_j$$



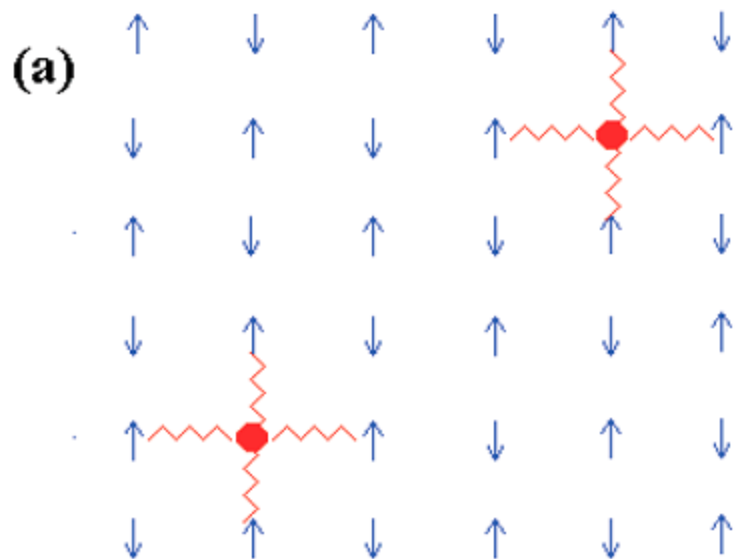
Spin flip breaks 6 bonds, costs 3J.



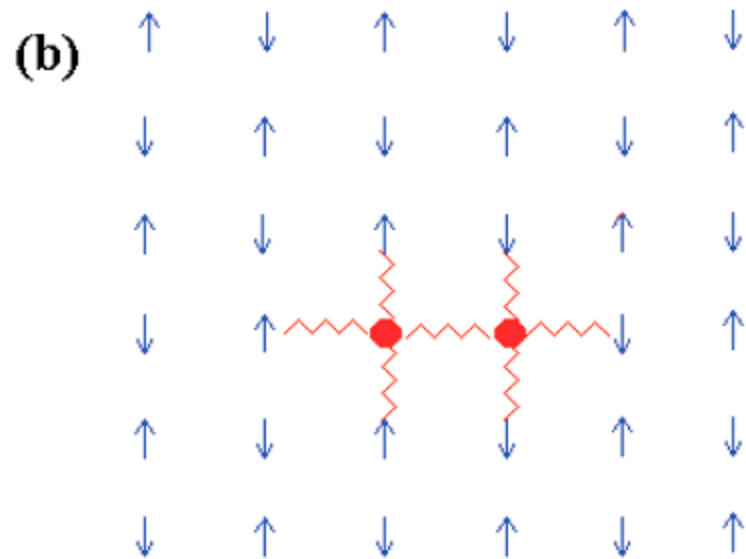
Largest J known among transition metal oxide, except for the Cu-O chain compound where J=220meV.

- **2-d hole-pairing:**

- AF-coupling causes holes to form pairs.



unpaired: 8 bonds broken



paired: 7 bonds broken